

BMAT Chemistry Compilation 2009 - 2016



Equilibria

6 The following reaction is at equilibrium:

$$2R(g) + 3S(g) \rightleftharpoons 2T(g)$$
 $\triangle H$ is negative

which conditions (A-E) would produce most of the product T?

	Temperature	Pressure	R	S	Catalyst
Α	high	low	add	remove	present
В	low	high	add	add	absent
С	low	low	add	add	absent
D	high	high	remove	add	present
E	low	high	remove	remove	absent

2013